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Oral iron and the bioavailability of zinc

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Abstract

The oral bioavailability of zinc was studied in non-pregnant adults before and 24 hours after two weeks of oral supplementation with iron and folic acid. Bioavailability was greatly reduced, and the shape of the plasma curves suggested that this was due to impairment of the intestinal absorption of zinc.

The findings suggest that the reduced bioavailability of zinc occurs because of interelement competition in the bowel wall. This might induce zinc depletion.

Introduction

We have reported that depletion of zinc in maternal tissues is strongly associated with depletion of zinc in fetal tissues and with idiopathic retardation of intrauterine growth.¹ The mechanisms responsible for maternal depletion are unknown, but in our local population a major cause is probably poor dietary intake, particularly as available zinc is present chiefly in expensive, high protein foods. In addition, dietary metals compete for intestinal absorption,² and therefore routine iron supple-

ments during pregnancy may decrease the absorption of zinc from the gut. Indeed, Solomons and Jacob³ reported that oral iron given concurrently with zinc to non-pregnant adults results in impaired oral bioavailability (or area under the plasma concentration time curve) of zinc. More prolonged effect of iron, however, has not been assessed.

We have studied the oral bioavailability of zinc given before and after oral supplementation with iron and folic acid.

Patients and methods

After a 12 hour fast 10 subjects (five men) were given 50 mg elemental zinc (220 mg zinc sulphate capsules) by mouth. Two baseline 5 ml venous blood samples were taken through a heparinised butterfly cannula before dosing, and then at 30 minute intervals for six hours. The blood was transferred into plastic heparinised tubes, the plasma separated at 400 g for 20 minutes at 20°C, and two 1 ml aliquots of plasma—each diluted with 9 ml 1.0M HCl—assayed for zinc concentration by atomic absorption spectrophotometry (model IL 257, Instrumentation Laboratories).

Each subject was then given oral iron and folic acid supplements as one Pregaday tablet (ferrous fumarate (100 mg iron) and folic acid 350 µg) daily for 14 days and zinc bioavailability reassessed 24 hours after the last dose.

In two healthy adults the reproducibility of the areas under the plasma concentration time curves in response to the oral administration of zinc sulphate (220 mg) was studied five times at two week intervals.

The elimination rate constant was calculated by least squares regression analysis of the log plasma zinc concentrations from 30 minutes after the peak until return to the baseline. The time to peak, peak plasma concentration, area under the curve, and elimination rate constant were compared before and after supplementation by two tailed Student's *t* test.

All patients gave written, informed consent and the study was approved by the hospital's ethical committee.

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Results

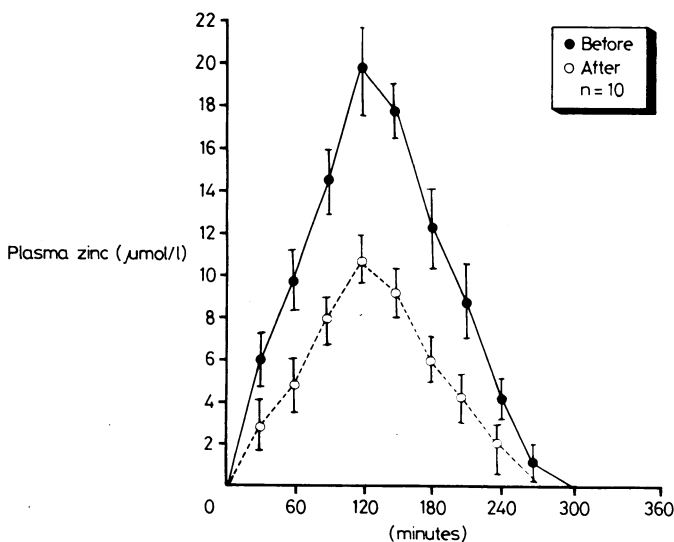
The area under the curve after oral zinc did not change with time, and the coefficient of variation was 9.5%. The mean peak plasma zinc concentration, area under the curve, and elimination rate constant were significantly lower, smaller, and slower respectively after two weeks of iron and folic acid supplementation (table). The times to achieve the peak concentration and return to baseline values, however, were unchanged (figure).

Plasma concentration time curve after oral zinc sulphate 220 mg before and after iron and folic acid supplementation. Values are means (SEM in parentheses)

	Area under curve ($\mu\text{mol h/l}$)	Time to peak (min)	Peak concentration ($\mu\text{mol/l}$)	Elimination rate constant/hour
Before supplementation	47.9 (3.8)	129.0 (1.2)	37.7 (2.4)	0.36 (0.036)
After supplementation	22.8 (4.0)	116.0 (3.1)	28.9 (1.7)	0.24 (0.036)
Significance of difference	$p < 0.01$	NS	$p < 0.05$	$p < 0.01$

NS = Not significant.

Conversion: SI to traditional units—Zinc: $1 \mu\text{mol/l} \approx 6.5 \mu\text{g}/100 \text{ ml}$.



Plasma concentration time curve responses before and after iron and folic acid supplementation, corrected for baseline values. Points are means; bars are SEMs.

Conversion: SI to traditional units—Zinc: $1 \mu\text{mol/l} \approx 6.5 \mu\text{g}/100 \text{ ml}$.

Discussion

Our study showed decreased oral bioavailability of zinc after iron and folic acid supplementation for two weeks in healthy non-pregnant adults. The result was not an effect of simultaneous administration of the two elements,³ for the zinc load was given 24 hours after the last iron supplement. Although we used pharmacological doses of zinc to define accurately the plasma response, a meal high in zinc content would contain at least half as much available zinc.

The most likely explanation for the decreased area under the curve and peak plasma zinc concentration and the slower elimination rate was that the intestinal absorption of zinc was impaired by the iron-folic acid tablets, and presumably by the iron rather than the folic acid, because of interelement competition.⁴ Alternative explanations were, however, considered. Supplementation with iron might reduce the binding of zinc to plasma proteins, increasing the concentration of free zinc and thereby leading to increased systemic clearance. This would lower peak concentrations and reduce the area under the curve, but the elimination rate constant would then be faster. If the volume of distribution of zinc increased owing to the greater availability of zinc for tissue binding lower peak plasma con-

centrations would occur but the area under the curve would not be affected. An increase in both clearance and volume of distribution of zinc would delay the return to baseline values, which was not observed.

Iron in the mucosa, however, by competing with zinc in the upper small intestine, will displace a larger mass of zinc down the intestine and delay its absorption. Such an effect would reduce the peak plasma concentration and area under the curve and, because of prolonged absorption, slow the apparent elimination rate, as was observed. Our results show that this effect lasts for at least 24 hours after iron ingestion.

Normally the intracellular metal binding protein metallothionein in the intestinal mucosa and liver probably regulates the amount of zinc being absorbed and reaching other tissues.⁵ Iron, however, does not bind to this protein, and so we cannot explain this competition, although zinc has been shown to impair iron absorption.³

We have previously shown that the zinc content of maternal leucocytes in pregnancy is strongly associated with idiopathic intrauterine growth retardation,¹ and suggested that, because many of our mothers are from the lower social classes, deprivation of dietary zinc from expensive, high protein food was probably important in their tissue zinc depletion. Hambidge *et al*⁶ found that plasma zinc concentrations in pregnancy were inversely correlated with the daily dose of oral iron supplementation. This and our results in non-pregnant adults suggest that oral iron supplementation might induce zinc depletion. Impairment of zinc absorption—particularly when the diet is poor and requirements are increased, as in pregnancy—casts further doubt on the wisdom of unselected iron supplementation in pregnancy,⁷ but we cannot yet recommend widespread supplementation with zinc in pregnancy. Further work is needed to determine whether these results are applicable to smaller doses of zinc in pregnancy.

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ONE HUNDRED YEARS AGO The following story, told in connection with a terrible outbreak of cholera on board an emigrant steamer, illustrates a remarkable degree of confidence in the profession. The deaths on board were so numerous that, for the purpose of indicating to the boatswain employed in removing the dead for burial, a cross was chalked against the berths where the corpses lay. By mistake, the fatal mark was attached to a berth where the inmate was engaged in a sound sleep. On the boatswain seizing him by the legs for removal, the intended victim called out loudly that he was not dead, but asleep—a remonstrance which drew from the old Scotch sailor the rejoinder, "Tut, tut, mon; the doctor knows better than you"; and who forthwith proceeded to endeavour to execute his commission. (*British Medical Journal* 1883;i:1079-80.)